Safety & Environmental Assurance Centre



()nilever

Deriving fish and *Daphnia* toxicity QSARs for anionic surfactants by using Durham experimental and computational membrane-water partition coefficients University

Katarzyna Przybylak^a, Andrea Gredelj^a, Elin L. Barrett^a, Jayne Roberts^a, Eoin Kearney^b, Alexandre Teixeira^a, Nicola Haywood^a, Mark A. Miller^b, Geoff Hodges^a ^a Safety and Environmental Assurance Centre, Unilever, Bedford, MK44 1LQ, UK ^b Department of Chemistry, Durham University, South Road, Durham DH1 3LE, UK

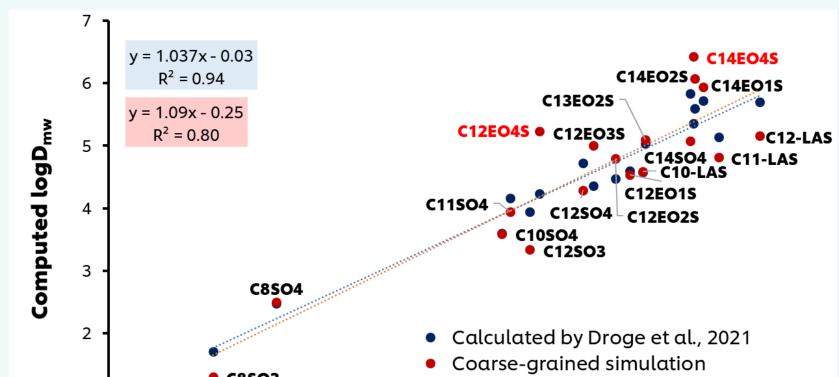
1. INTRODUCTION

- Quantitative Structure Activity Relationships (QSARs) are a viable alternative to *in-vivo* toxicity testing of chemicals.
- Many (eco)toxicity QSARs are hydrophobicity-based relationships using the octanol-water partition coefficient, logK_{ow} as a common descriptor for chemicals' toxicity. Determination of logK_{ow} for ionisable chemicals and surfactants is empirically difficult due to their tendency to accumulate at the octanolwater interface. Predictive methods are also often unreliable [1] Octanol cannot adequately describe the interactions of polar, charged, or amphiphilic compounds within ordered 3D structures of biological membranes. Membrane-water partition or distribution coefficient (logK_{mw} or **logD**_{mw}) provides a more biologically realistic approach for these compound types. As experimental determination of $logK_{ow}/D_{mw}$ can also be complex and time consuming, we have calculated $\log K_{mw} / D_{mw}$ using the regression method developed by Droge et al. [2] and coarse-grained simulations [3] to develop **fish and** *Daphnia* **toxicity QSARs for anionic surfactants**.

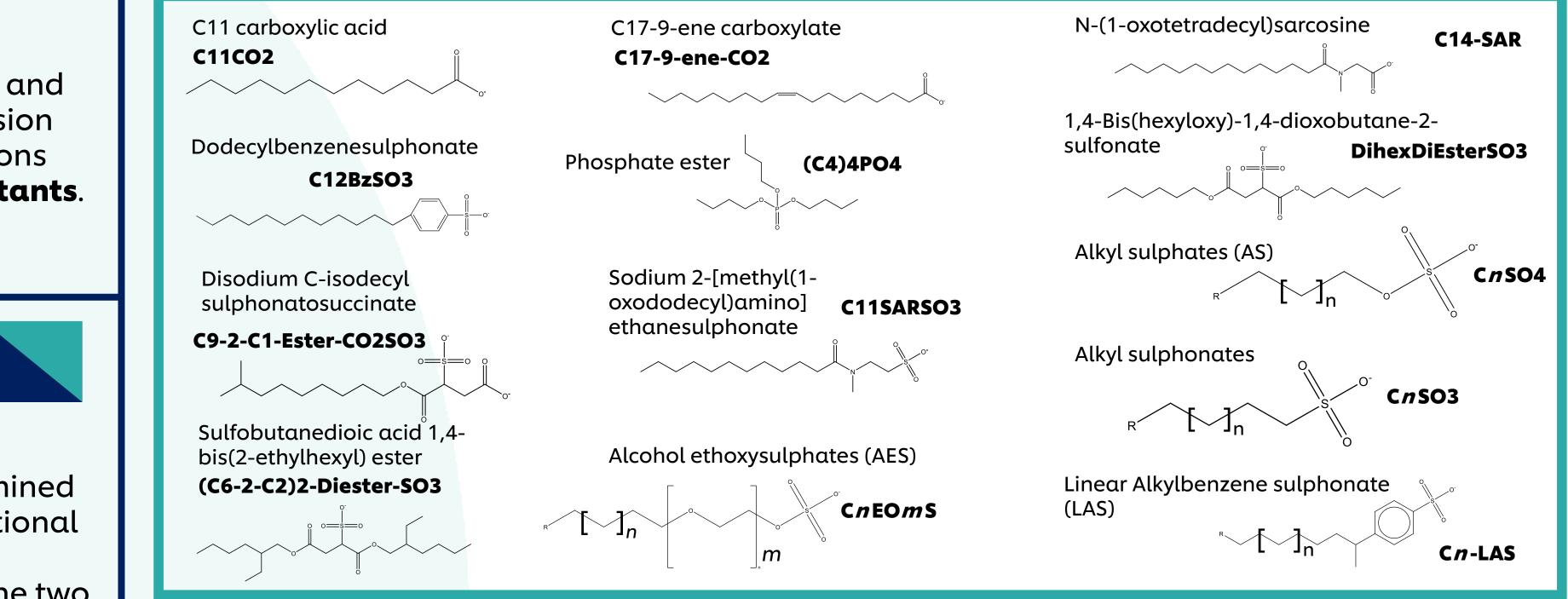
2. METHODOLOGY

- Previously, we have shown good correlation of simulated logD_{mw} against experimental values of chemicals, and presented several QSARs with homologue series of anionic surfactants successfully using $log D_{mw}$ as an (eco)toxicity proxy [4,5].
- Here, database of **fish and** *Daphnia* literature toxicity data [6-9] containing **mono** constituent anionic surfactants (Table 1) was compiled and used to develop new QSARs. To address logD_{mw} experimental data gaps regarding the chemical space coverage (limited surfactant groups) we are also using two computational methods for logD_{mw} (Droge et al. [2] & coarse-grained simulation [3]), with the advantages and disadvantages of various methods for deriving logD_{mw} being discussed. Finally, we have used QSARs based on the same chemical space to make preliminary comparisons of species sensitivity for fish and Daphnia.

3. LogD_{MW} METHODS EVALUATION



Comparison of experimentally determined logD_{mw} with computational data indicates good correlation between the two. Table 1. List of used anionic surfactants (with chemical structures and abbreviations)



• The Droge et al. method is a multiple regression based equation with limitations in surfactant group coverage (e.g. sarcosinates, phosphate esters), whereas simulation method is not restricted to certain surfactant groups.



However, we have identified a limitation in simulation method for chemicals containing ethoxylate units (EO > 4).



Figure 1. Comparison of calculated $log D_{mw}$ with measured $log D_{mw}$ (SSLM method from Droge, 2019 [10] & Droge et al., 2021 [3] & liposome internal data from SEAC, Unilever)

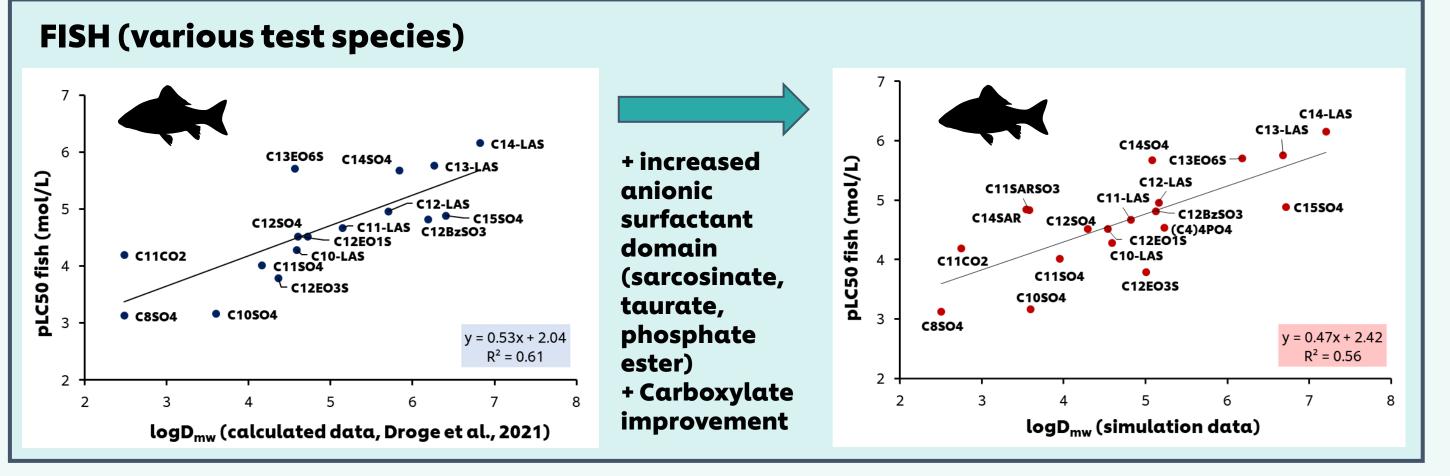


Figure 2. Comparison of fish QSARs based on calculated logD_{mw} by the Droge et al. regression and coarse-grained simulations

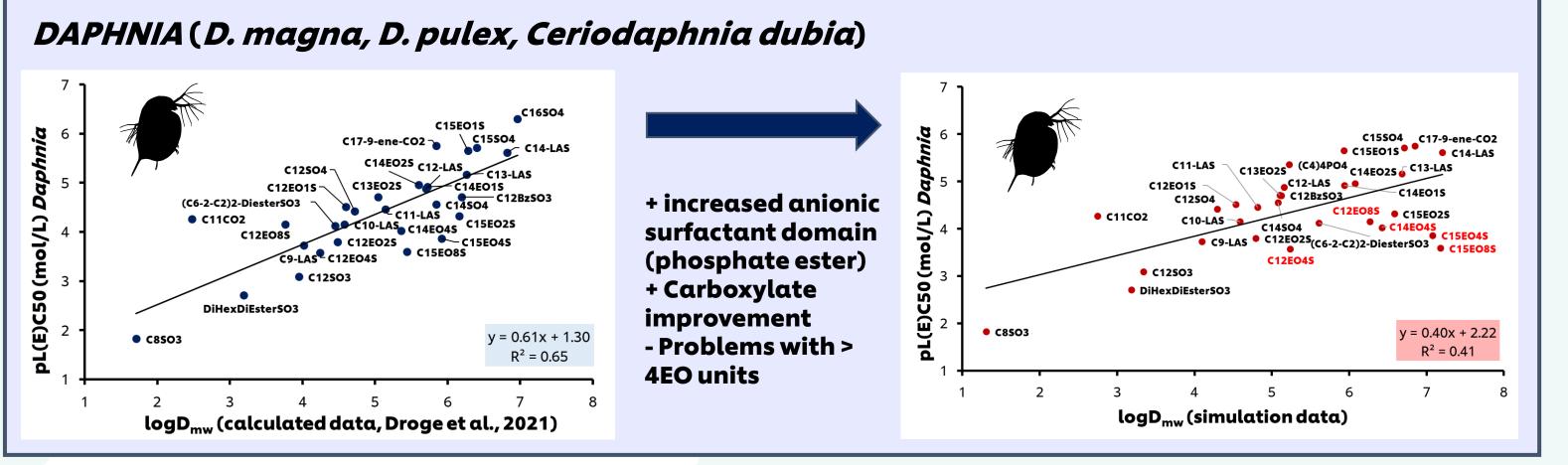
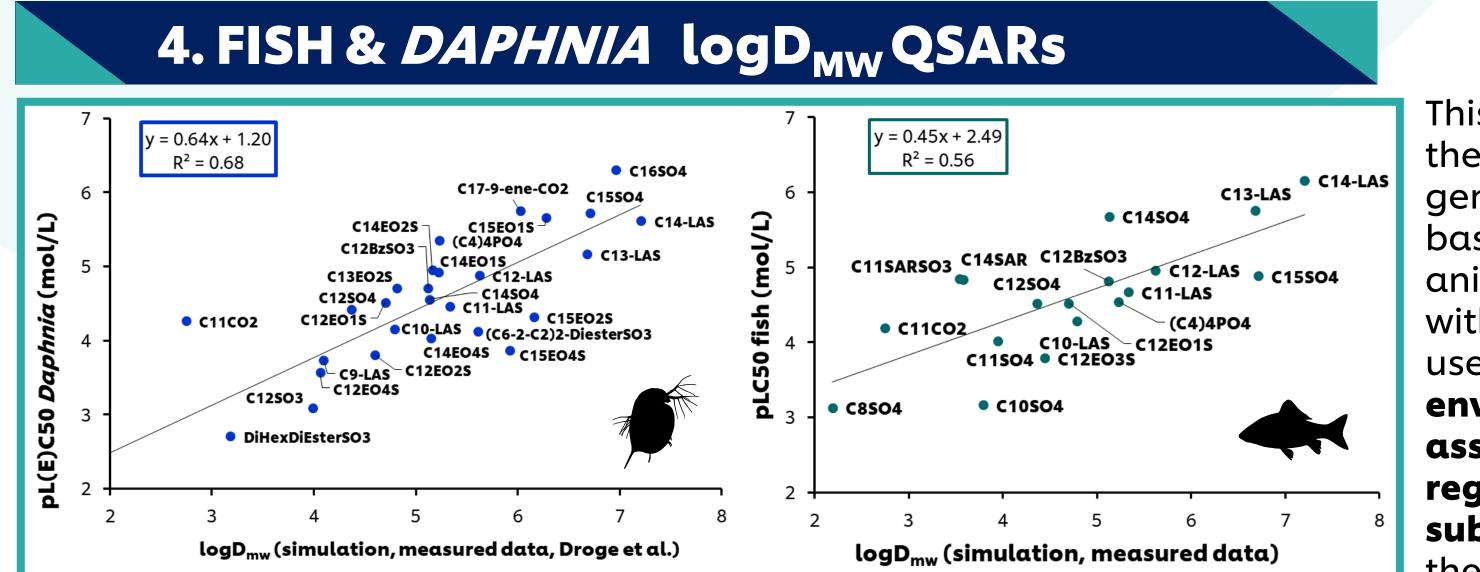


Figure 3. Comparison of Daphnia QSARs based on calculated logD_{mw} by the Droge et al. regression and coarse-grained simulations



This work represents the first step towards generating logD_{mw} based QSARs for anionic surfactants with potential to be used in environmental risk

5. CONCLUSIONS AND FUTURE WORK

- Using literature (eco)toxicity data for a variety of mono constituent anionic surfactants we have shown it possible to generate general anionic surfactant - based QSAR using logD_{mw} for the prediction of aquatic toxicity to fish and Daphnia.
- **Species sensitivity** between fish and *Daphnia* based on the same chemical dataset suggests that there is no difference between the two. This could provide the opportunity to waive the need for acute

Figure 4. Anionic surfactants fish and *Daphnia* QSARs, using a combination of applicable logD_{mw} methods

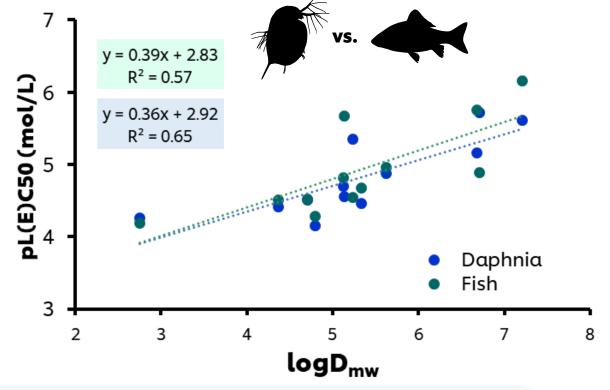


Figure 5. Fish and Daphnia QSARs comparison

assessment and/or regulatory submissions avoiding the need for unnecessary fish testing.

- QSARs were compared for chemicals where both Daphnia and fish toxicity data are available (including LAS, AS, AES, phosphate esters).
- Statistically significant differences were **not** observed between slopes and intercepts, indicating there is **potentially no** difference between sensitivity of these trophic levels to anionic surfactants.

fish toxicity testing for anionic surfactants. However, further work would be required to expand the data set to provide a more robust analysis.



Development of a reliable computational logD_{mw} method which covers a wide chemical space is required. Further work has been identified to **refine the approach for chemicals with EO** > **4**.



To improve the QSAR regression, consideration could be given to include additional (eco)toxicity data from multiconstituent surfactants, however this requires previous knowledge of the chain length distribution in order to predict a representative $log D_{mw}$.

6. REFERENCES

[1] G. Hodges et al., Environ Sci Eur **2019**, 31, 1 [2] S. Droge et al., Environ. Sci.: Processes Impacts, **2021**, 23, 1930 [3] T. D. Potter et al., J. Chem. Theory Comput. **2021**, 17, 9, 5777–5791 [4] A. Gredelj *et al.*, SETAC EU 2022, ISSN 2310-3043 [5] E. Barrett *et al.*, SETAC SciCon NA 2021, ISSN 1087-8939 [6] USEPA EECOTOX database (ECOTOX | Home (epa.gov)),

Fish 96h LC50, Daphnia 48h L(E)C50 [15/02/2023] [7] eChemportal/ECHA REACH database (eChemPortal) [15/02/2023] [8] G. Hodges et al., Chemosphere **2006**, 63, 1443-1450 [9] S. Dyer et al., Environ. Toxicol. Chem **2000**, 19, 3, 608-616 [10] S. Droge, Environ. Sci. Technol., 2019, 53, 760-770